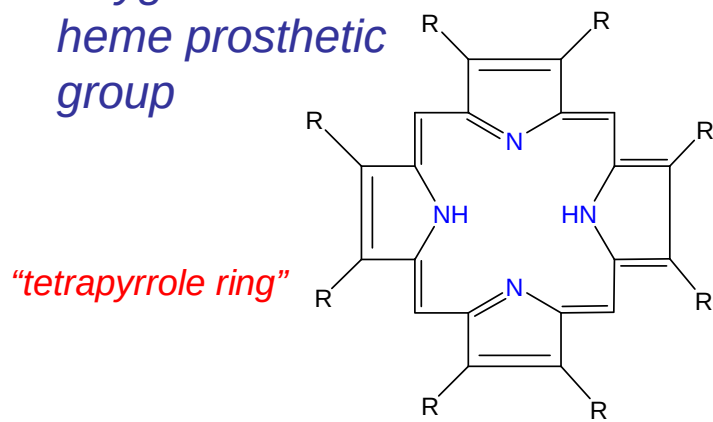


Protein conformation, dynamics and function

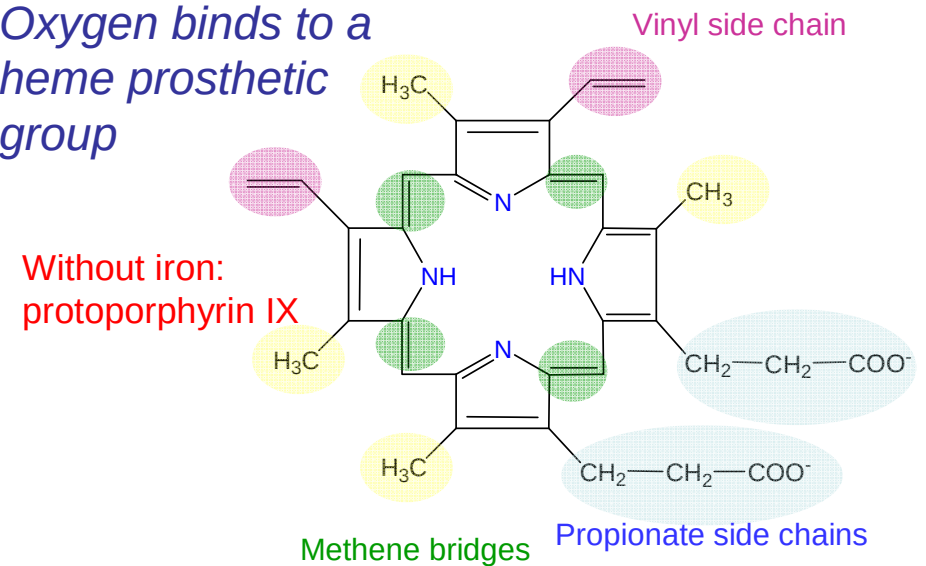
Key principles:

- Protein function involves binding of other molecules
“Ligand”
- Proteins contain ligand “binding sites”
 - highly specific
 - single or many sites
 - 1 or more different ligand type
- Proteins are not rigid “conformational change”
- Conformational changes are often linked to ligand binding “induced fit”
- Conf. change, site 1 $\leftrightarrow$ conf. change, site 2
- Protein-ligand interactions may be regulated

Oxygen binds to a
heme prosthetic
group

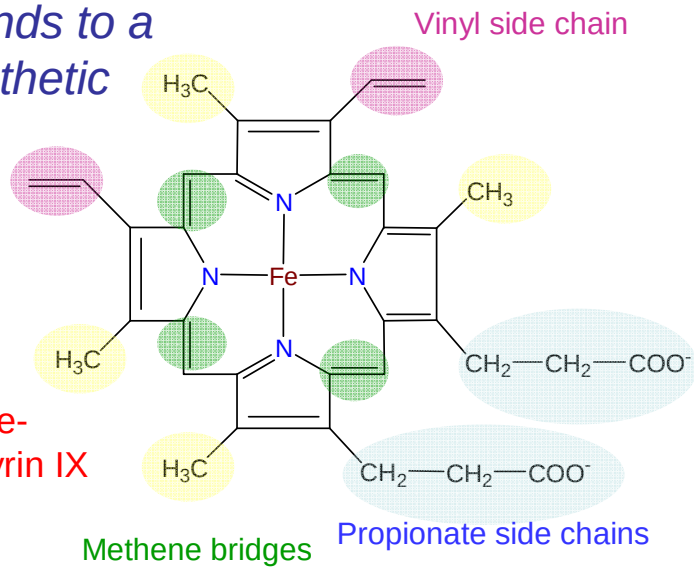


Oxygen binds to a
heme prosthetic
group

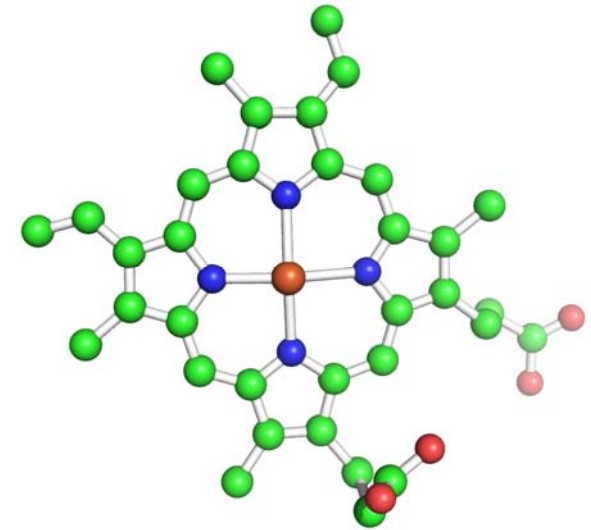


Oxygen binds to a heme prosthetic group

With iron: Fe-protoporphyrin IX (heme)



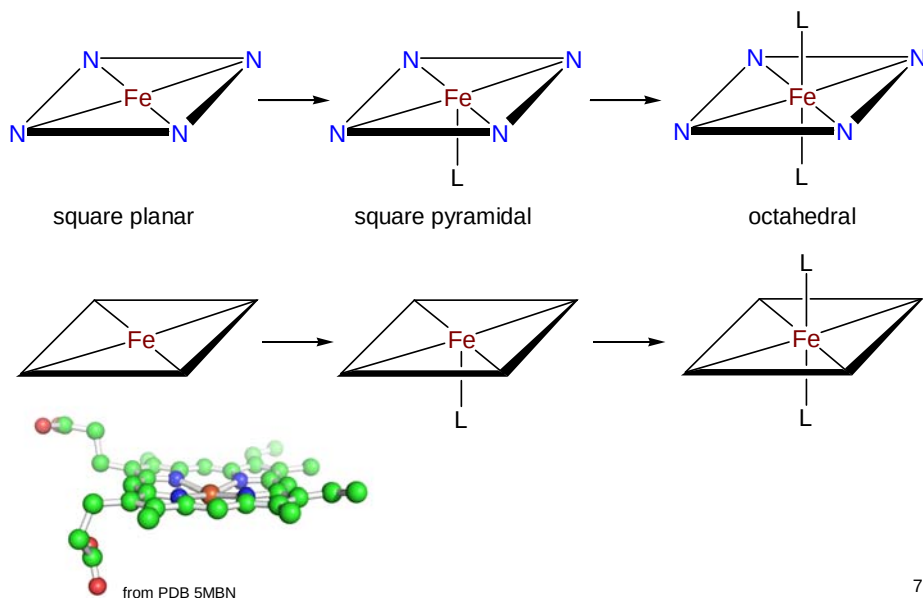
5
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from PDB 5MBN

6
© S. Ensign

5th and 6th coordination sites on iron



7
© S. Ensign

Myoglobin

- O₂ storage protein
- 3-D structure: globin fold, 75% α -helix
- Single polypeptide, Mr = 16,700, 153 AA
- 1 heme/polypeptide
- Interior is hydrophobic except for 2 his residues

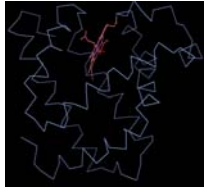


Lopez.kin

8
© S. Ensign

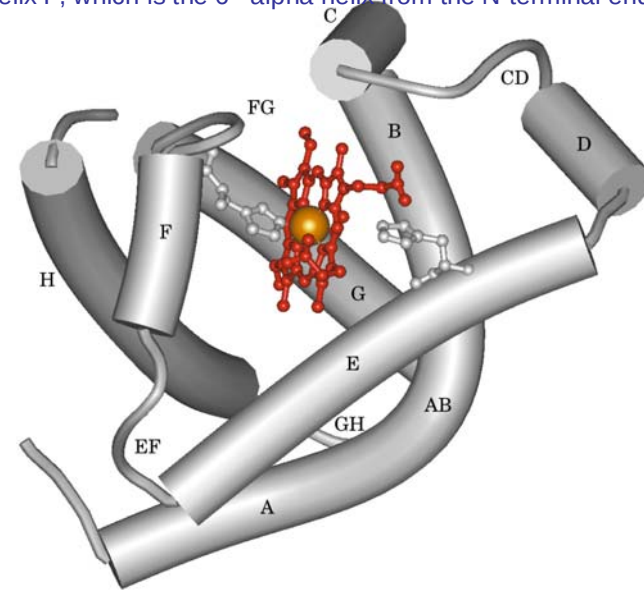
Myoglobin

- O₂ storage protein
- 3-D structure: globin fold, 75% α -helix
- Single polypeptide, Mr = 16,700, 153 AA
- 1 heme/polypeptide
- Interior is hydrophobic except for 2 his residues

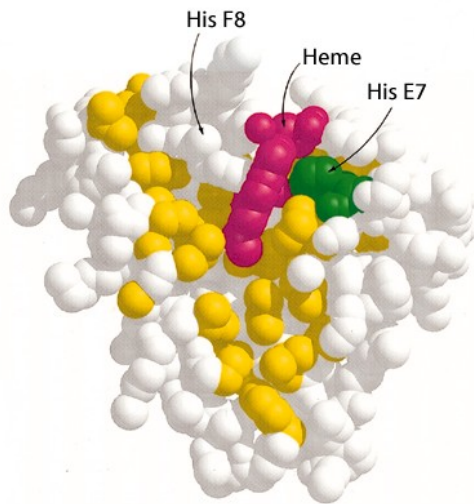


Lopez.kin

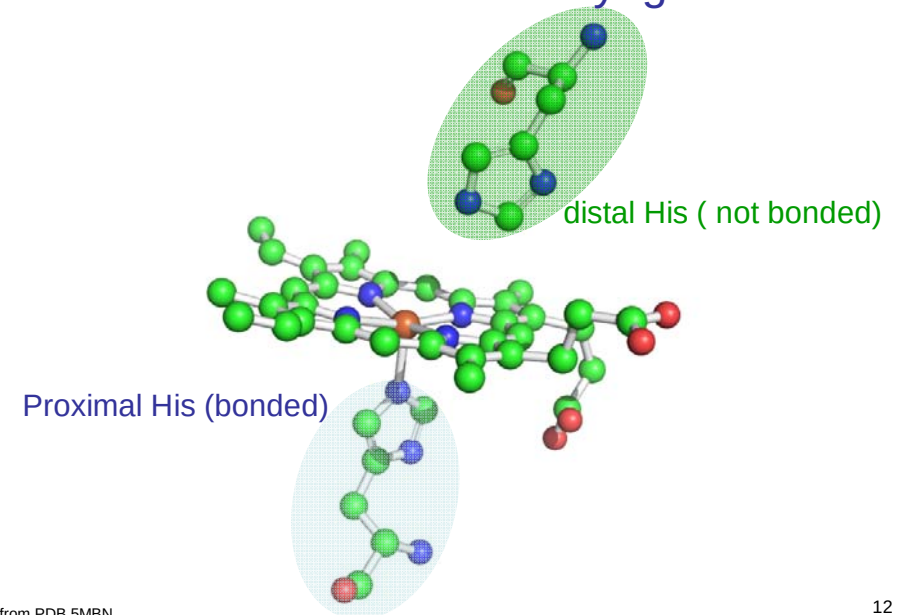
Note helix designations using letters. Residue "F8" is the eighth residue in helix F, which is the 6th alpha helix from the N-terminal end



Note helix designations using letters. Residue "F8" is the eighth residue in helix F, which is the 6th alpha helix from the N-terminal end



Heme environment in myoglobin



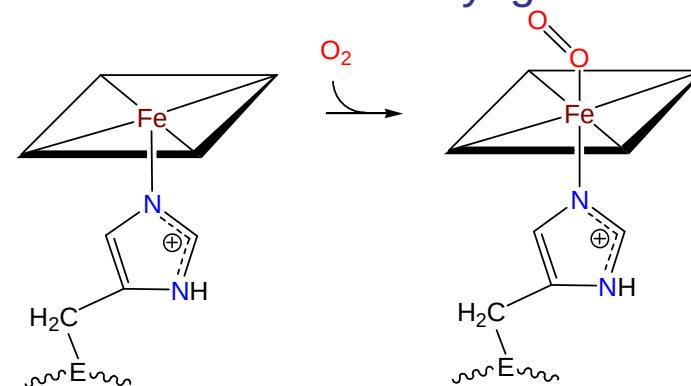
The occupation status of the 6th coordination site is critical to the physiological status of myoglobin and hemoglobin

Form	Oxidation state	Ligand at 6 th position
Deoxy Mb	Fe ²⁺	None
Oxy Mb	Fe ²⁺	O ₂
Ferri Mb	Fe ³⁺	H ₂ O

O₂ can only bind ferrous heme!

13
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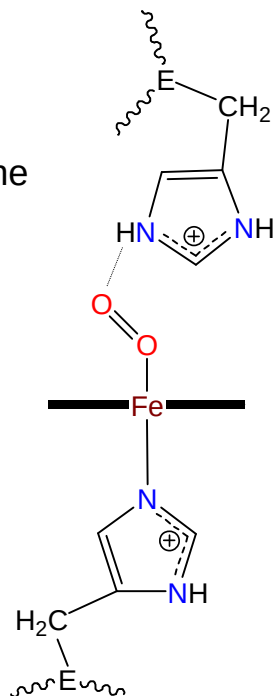
Binding of oxygen to the 6th coordination site of heme in myoglobin



O₂ is sp² hybridized, and thus binds to Fe at a 120 degree angle

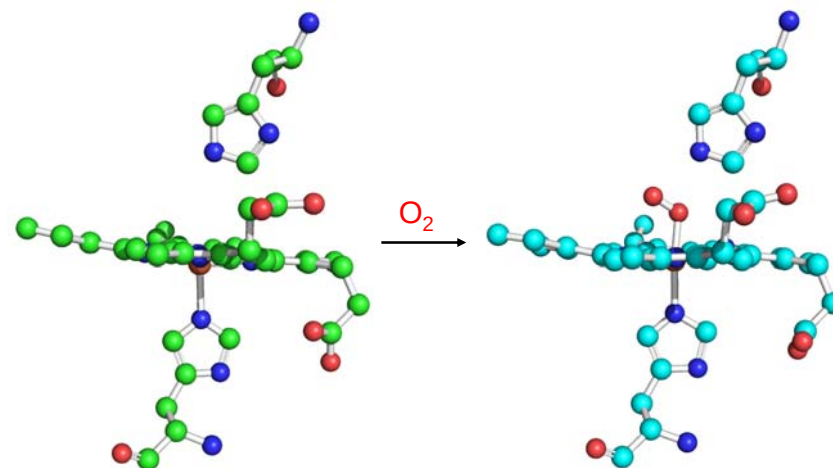
14
© S. Ensign

O₂ is sp² hybridized, and thus binds to Fe at a 120 degree angle, which is facilitated by the position of the distal histidine



15
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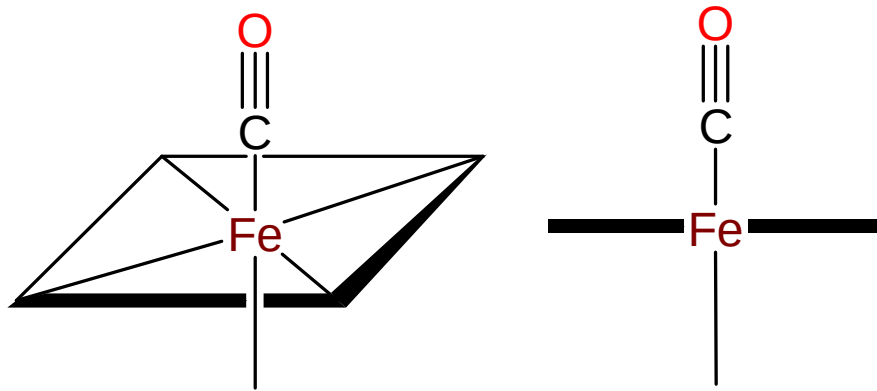
Binding of oxygen to the 6th coordination site of heme in myoglobin



from PDB 5MBN

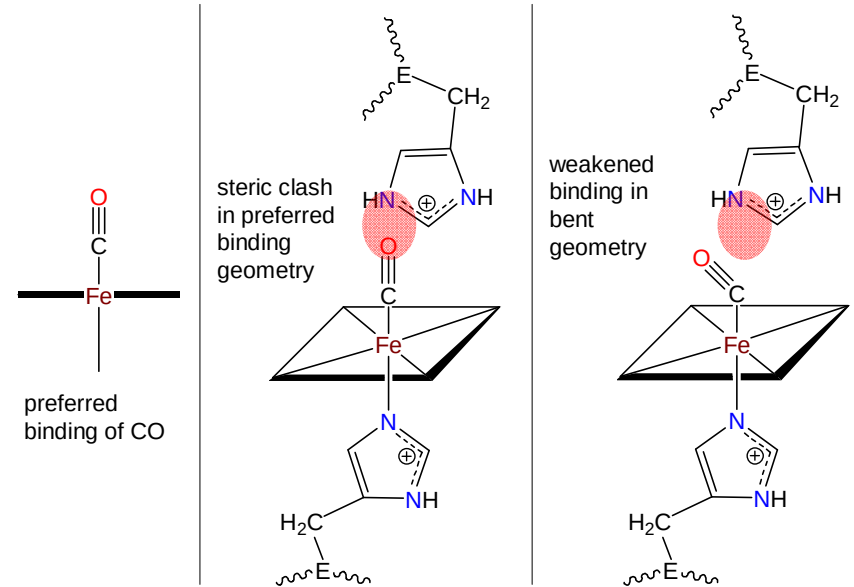
from PDB 1MBQ
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CO has a high affinity for the iron in heme



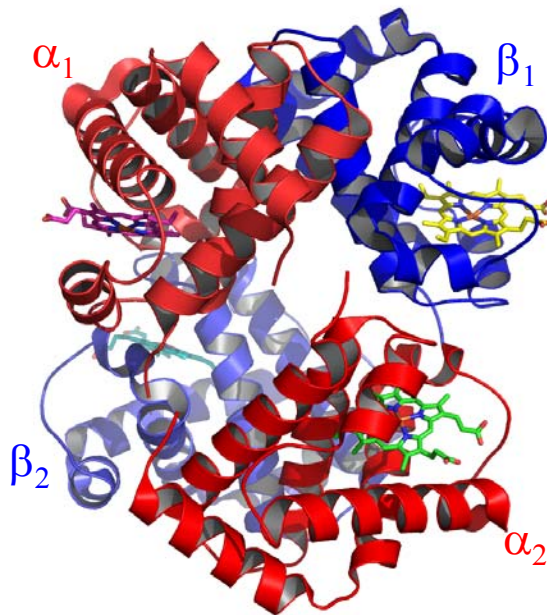
17
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The distal His prevents linear binding of CO, which is preferred for the sp hybridized CO molecule



18
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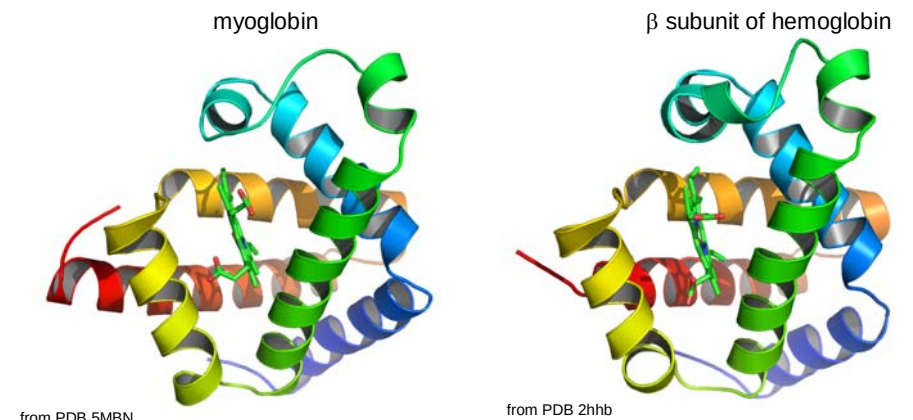
Hemoglobin:
 $\alpha_2\beta_2$ tetramer
 O_2 transport



from PDB 2hhb

19
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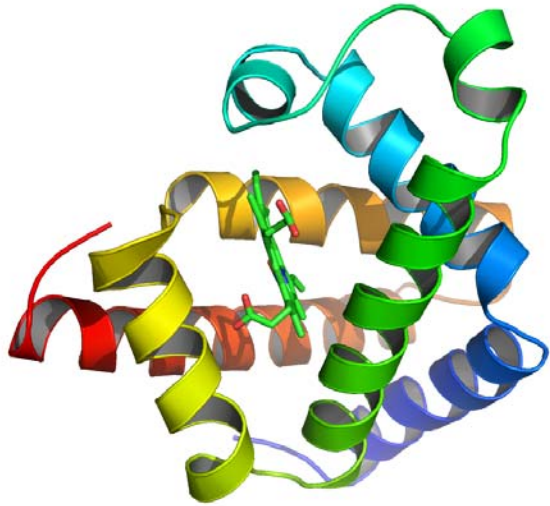
Myoglobin and the subunits of hemoglobin are built from the same structural motifs



AA sequences are only "identical" at 24 of 141 positions!

20
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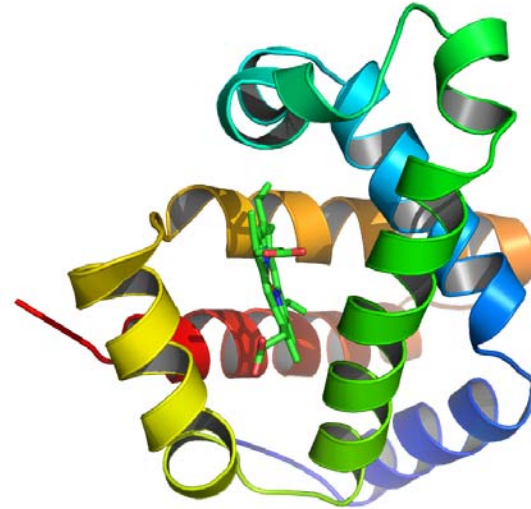
“Fold” structure of myoglobin



from PDB 5MBN

21
© S. Ensign

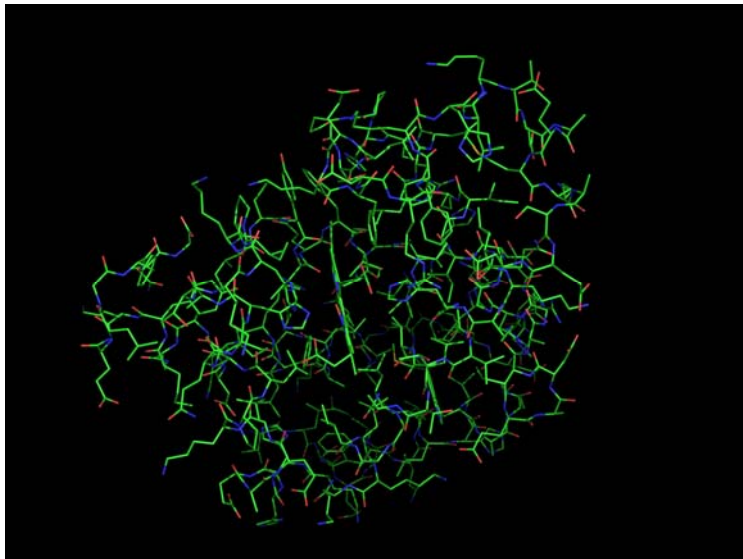
“Fold” structure of beta subunit of hemoglobin



from PDB 5hhb

22
© S. Ensign

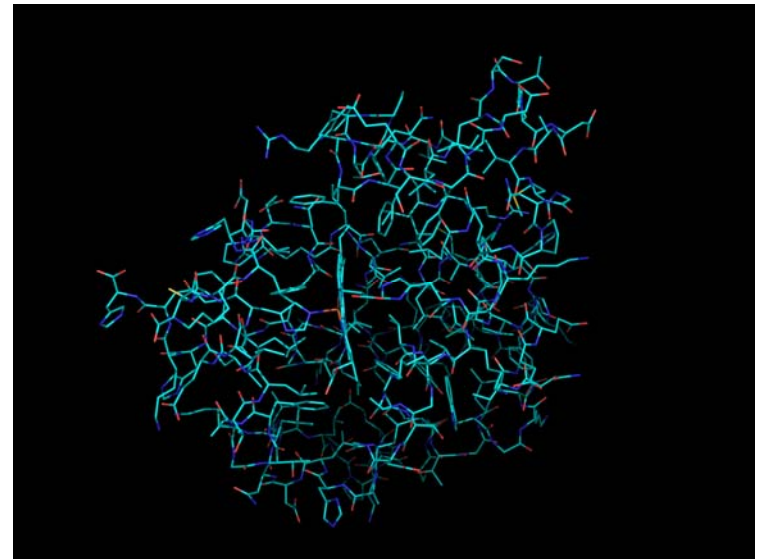
Complete structure of myoglobin



from PDB 5MBN

23
© S. Ensign

Complete structure of beta subunit of hemoglobin



from PDB2hhb

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Different amino acid sequences can specify similar 3-D structures

- Some residue positions are “highly conserved” (identical)
 - His93, His64, 7 others
- Some residue positions allow conservative changes (similar AA)
 - Nonpolar residues in interior (ala → ile)
- Some residue positions are not conserved (high variability)
 - Hydrophilic surface residues

25
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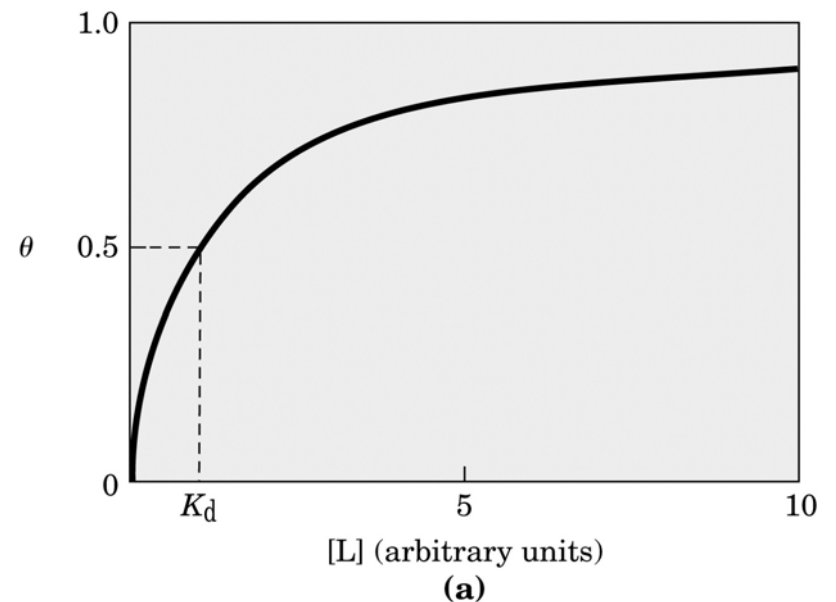
Allosteric interactions: interactions between spatially distinct sites

- Hb is an allosteric protein, Mb is not
- Binding of O₂ to one site on Hb enhances binding of additional O₂ to other sites on the same molecule “cooperativity”
- Other molecules affect the affinity of O₂ for Hb:
 - pH
 - CO₂
 - 2,3-diphosphoglycerate

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Mathematical and graphical representations of ligand binding to proteins

A hyperbola describes the simple noncooperative binding of a ligand to a protein



27
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table 7-1 The lower the K_d , the tighter the binding

Some Protein Dissociation Constants

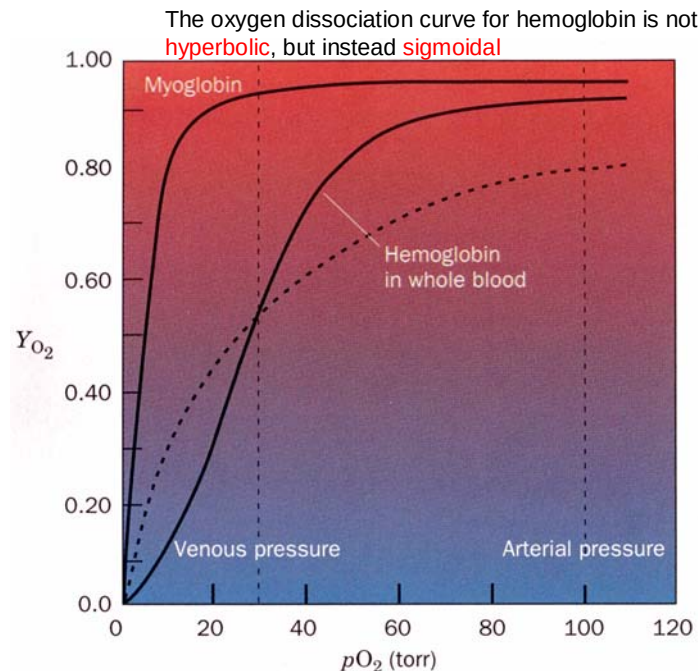
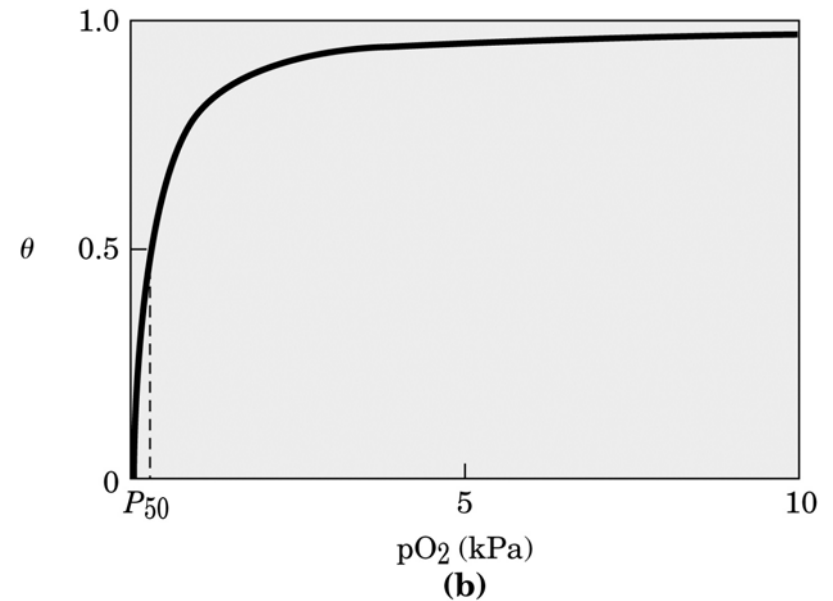
Protein	Ligand	K_d (M)*
Avidin (egg white) [†]	Biotin	1×10^{-15}
Insulin receptor (human)	Insulin	1×10^{-10}
Anti-HIV immunoglobulin (human) [‡]	gp41 (HIV-1 surface protein)	4×10^{-10}
Nickel-binding protein (<i>E. coli</i>)	Ni^{2+}	1×10^{-7}
Calmodulin (rat) [§]	Ca^{2+}	3×10^{-6}
		2×10^{-5}

*A reported dissociation constant is valid only for the particular solution conditions under which it was measured. K_d values for a protein-ligand interaction can be altered, sometimes by several orders of magnitude, by changes in solution salt concentration, pH, or other variables.

[†]Interaction of avidin with the enzymatic cofactor biotin is among the strongest noncovalent biochemical interactions known.

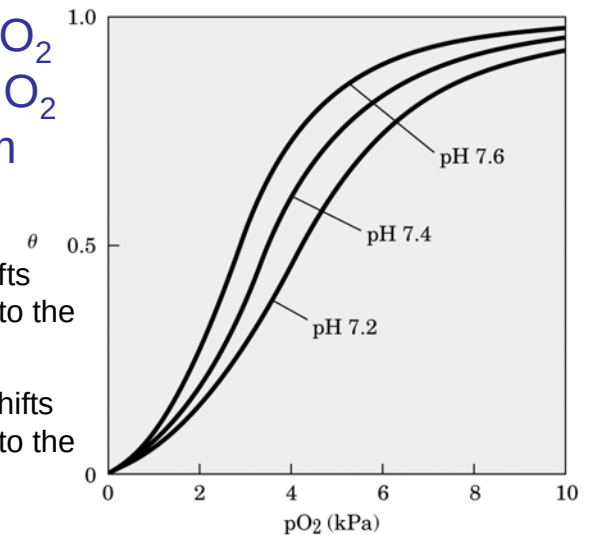
[‡]This immunoglobulin was isolated as part of an effort to develop a vaccine against HIV. Immunoglobulins (described later in the chapter) are highly variable, and the K_d reported here should not be considered characteristic of all immunoglobulins.

[§]Calmodulin has four binding sites for calcium. The values shown reflect the highest- and lowest-affinity binding sites observed in one set of measurements.



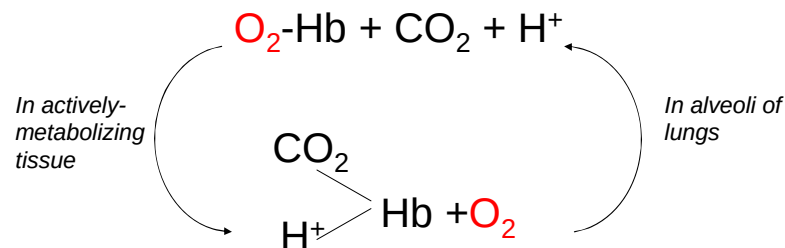
The Bohr effect:
Effects of CO_2 and $[\text{H}^+]$ on O_2 release from hemoglobin

- Increased $[\text{H}^+]$ shifts dissociation curve to the right
- Increased $[\text{CO}_2]$ shifts dissociation curve to the right



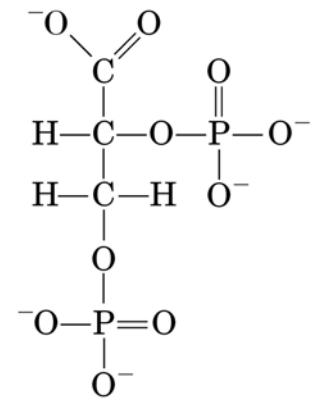
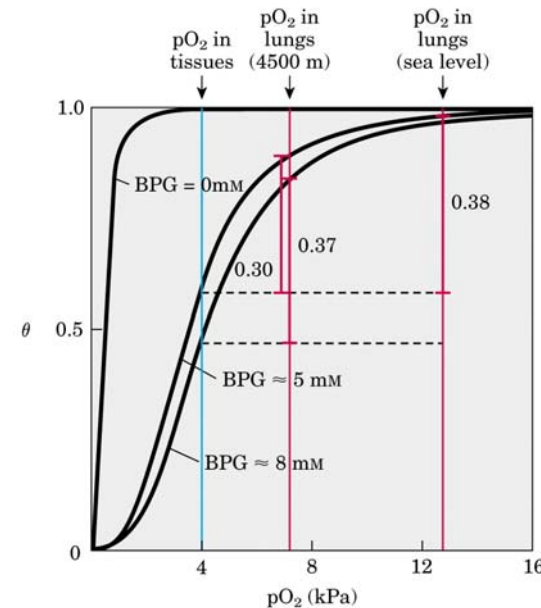
Increased CO_2 and H^+ concentrations lower O_2 affinity for Hb, i.e. promoting release of O_2 from oxyhemoglobin

Hemoglobin transports CO_2 and H^+ as well as O_2



33
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2,3-DPG lowers the oxygen affinity of Hb

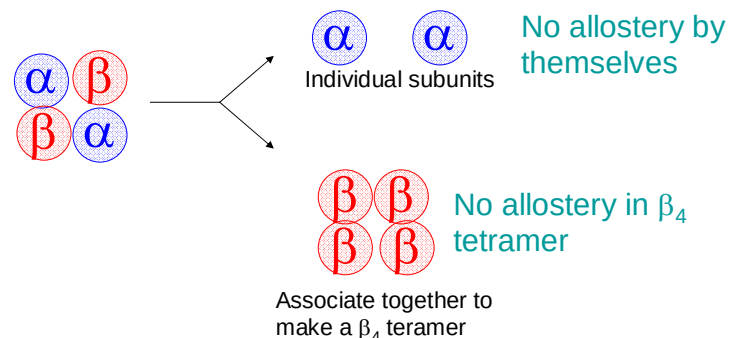


2,3-Bisphosphoglycerate

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What is the biochemical basis for allosteric effects in hemoglobin?

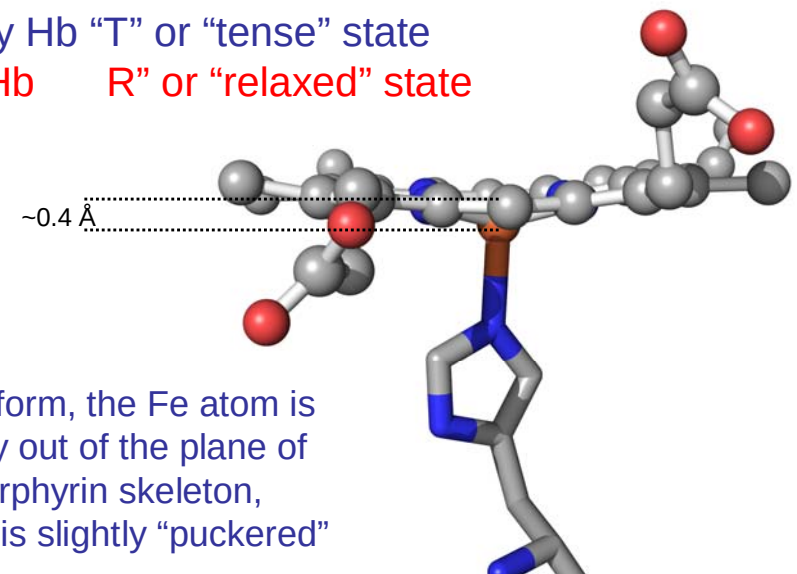
Expt. 1: Dissociate Hb into individual subunits



Conclusion: The allosteric properties of Hb must arise from interactions between subunits in the native quaternary structure

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Two conformations of Hb
Deoxy Hb "T" or "tense" state
Oxy Hb "R" or "relaxed" state



In "T" form, the Fe atom is slightly out of the plane of the porphyrin skeleton, which is slightly "puckered"

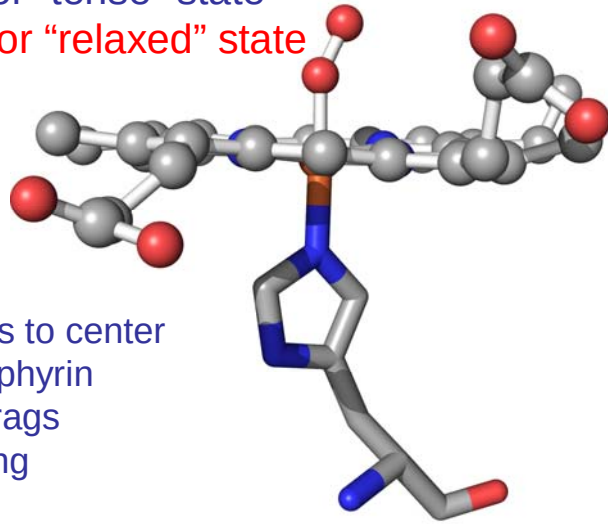
from PDB2hhb

36
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Two conformations of Hb

Deoxy Hb “T” or “tense” state

Oxy Hb “R” or “relaxed” state



T → R: Fe moves to center of porphyrin, porphyrin straightens out, drags proximal His along

from PDB1hhb

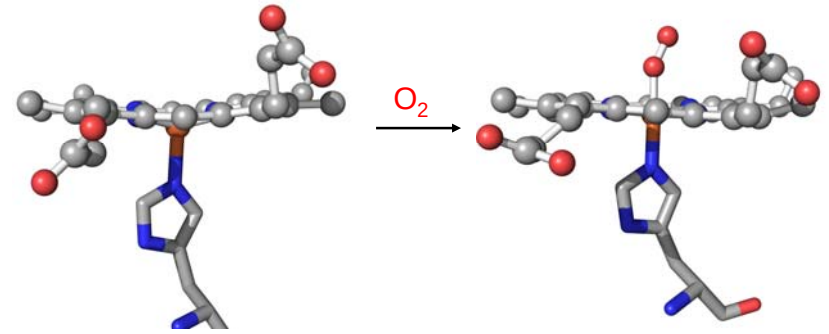
37
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Two conformations of Hb

•Deoxy Hb “T” or “tense” state

Oxy Hb “R” or “relaxed” state

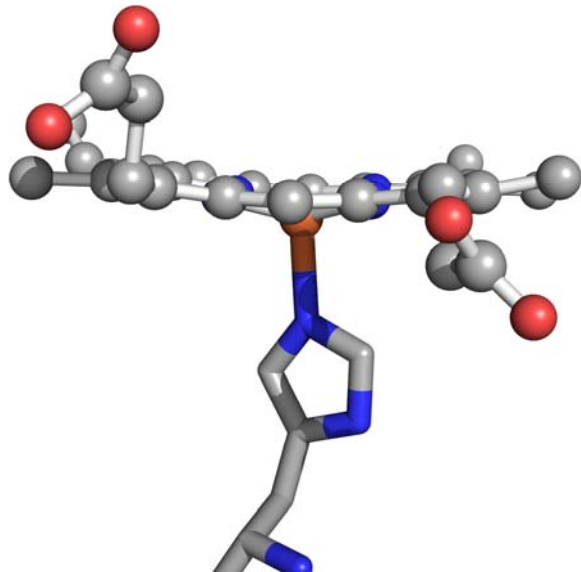
T → R: Fe moves to center of porphyrin, drags proximal His along, this tilts attached F helix



from PDB2hhb

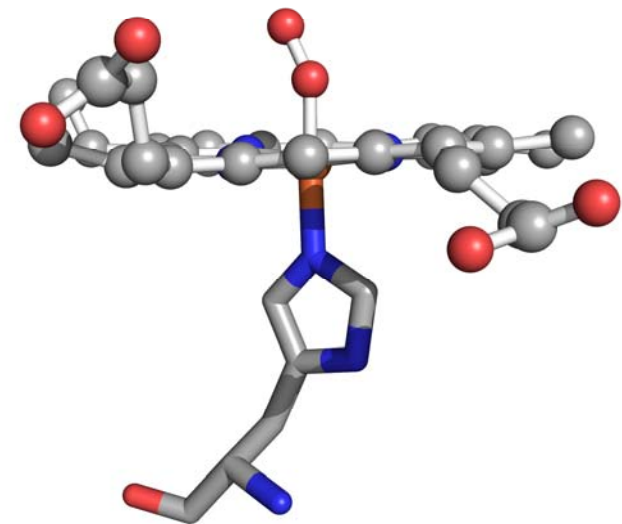
from PDB1hhb

38
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from PDB2hhb

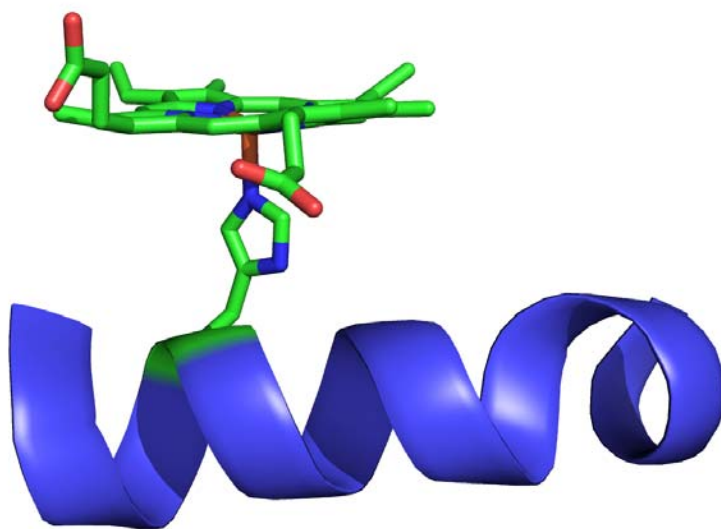
39
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from PDB1hhb

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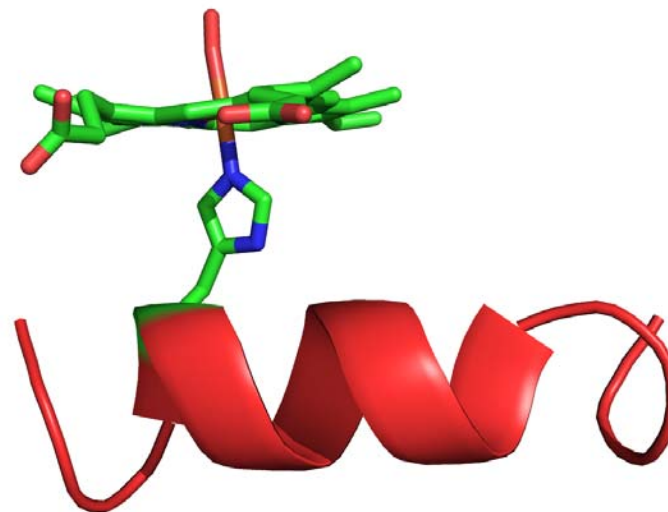
Note orientation of F helix with respect to heme in deoxyHb



from PDB2hhb

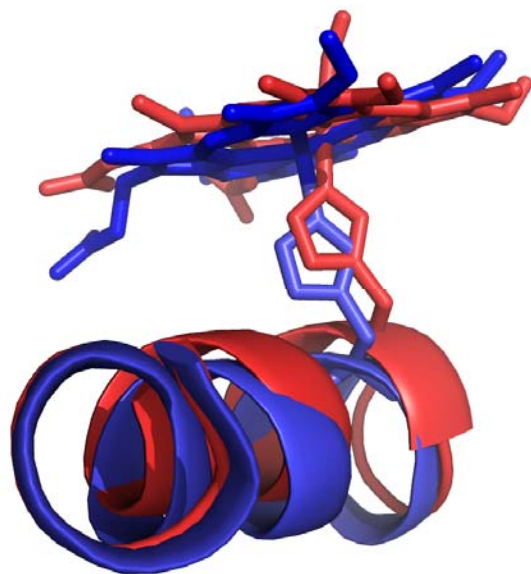
41
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Note orientation of F helix with respect to heme in oxyHb



from PDB1hho

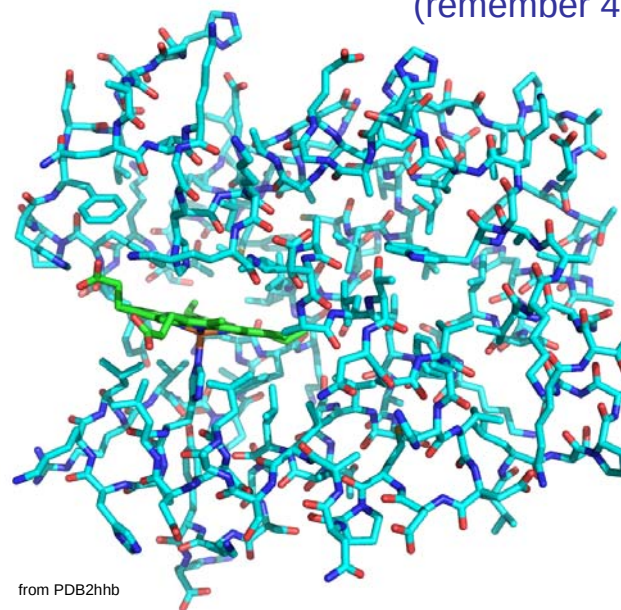
42
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from PDB2hhb and 1hho

43
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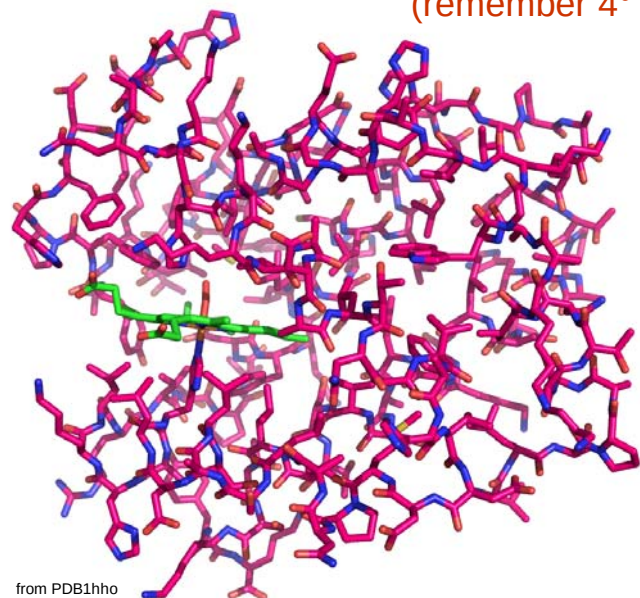
Conformation of a single alpha subunit in deoxyHb
(remember 4° structure is $\alpha_2\beta_2$)



from PDB2hhb

44
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Conformation of a single alpha subunit in oxyHb
(remember 4° structure is $\alpha_2\beta_2$)

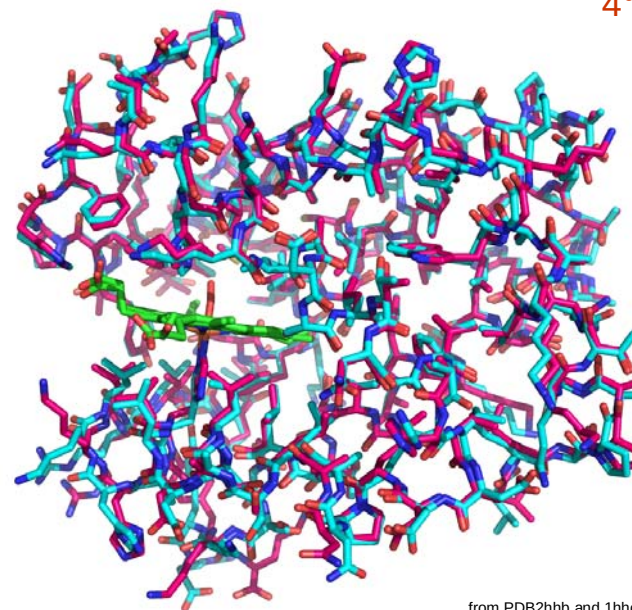


from PDB1hho

45

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Overlay of deoxy and oxyHb alpha subunits (remember
4° structure is $\alpha_2\beta_2$)

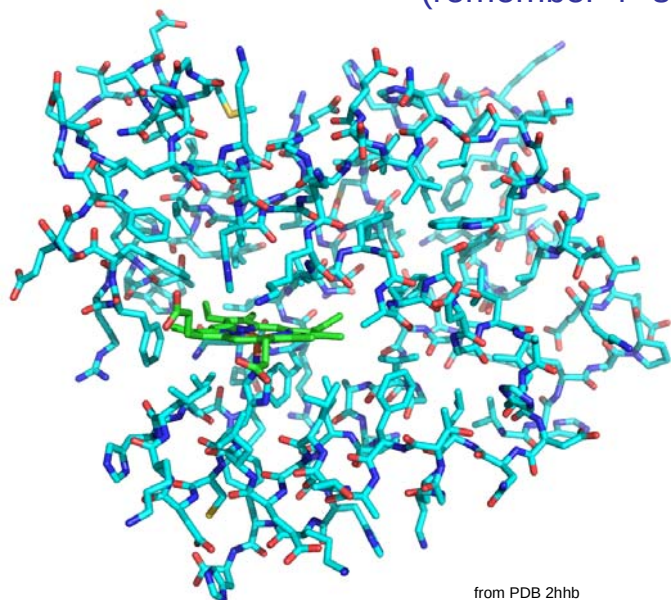


from PDB2hhb and 1hho

46

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Conformation of a single beta subunit in deoxyHb
(remember 4° structure is $\alpha_2\beta_2$)

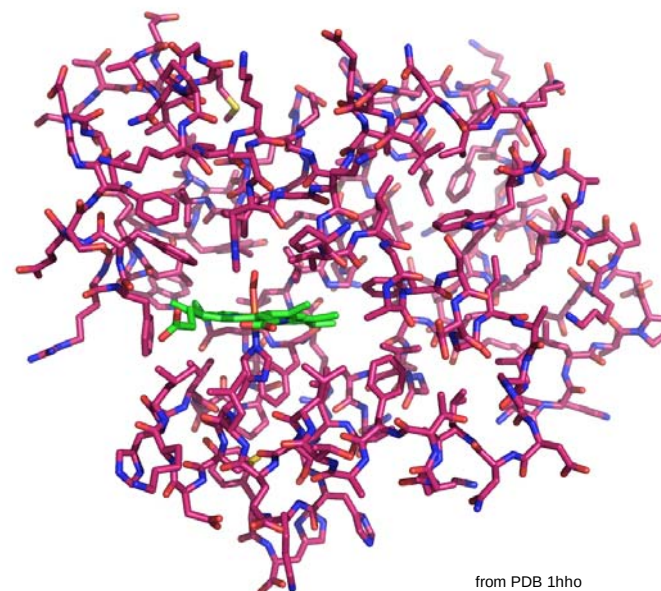


from PDB 2hhb

47

© S. Ensign

Conformation of a single beta subunit in oxyHb
(remember 4° structure is $\alpha_2\beta_2$)

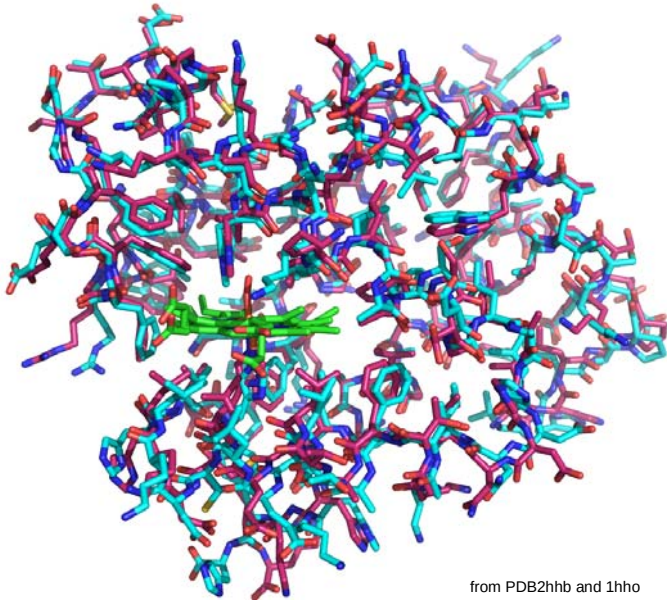


from PDB 1hho

48

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Overlay of deoxy and oxyHb beta subunits (remember
4° structure is $\alpha_2\beta_2$)



from PDB2hho and 1hho

49
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Summary of T and R hemoglobin structures

- T Hb (4° structure) is stabilized by multiple ionic interactions, H-bonds, and ion-dipole interactions within subunits (intrasubunit) or between subunits (intersubunit)
- O₂ binding to a “T” Hb subunit induces a conformational change in the heme, reorienting the F helix, which translates to an overall conformational change in the Hb subunit
- The change in conformation of the O₂-bound subunit leads to a reorientation/change of stabilizing interactions to adjacent subunits

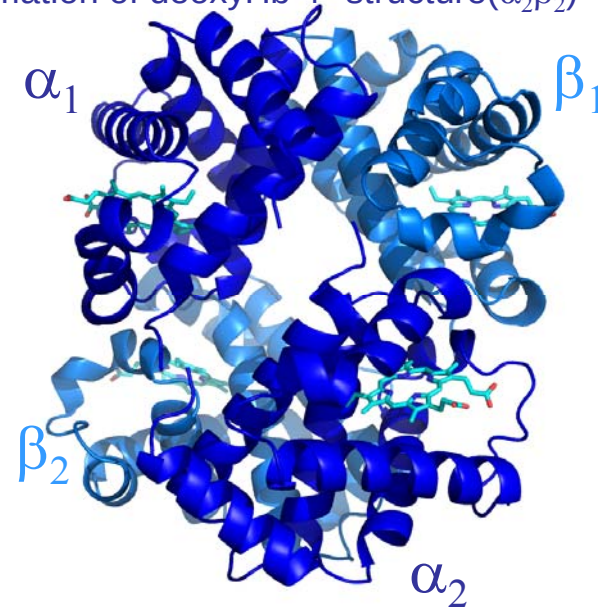
50
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Forces that stabilize T and R hemoglobin

- Intra and inter subunit interactions
- interactions between amino acids
 - ionic interactions: “salt bridges”
 - ionic attractions between carboxy and amino termini and charged side chains
 - ion-dipole attractions
 - H bonds
- Salt bridges between amino acids and modified (carbamylated) amino acids (T only)
- Salt bridges between amino acid residues and 2,3-diphosphoglycerate (T only)

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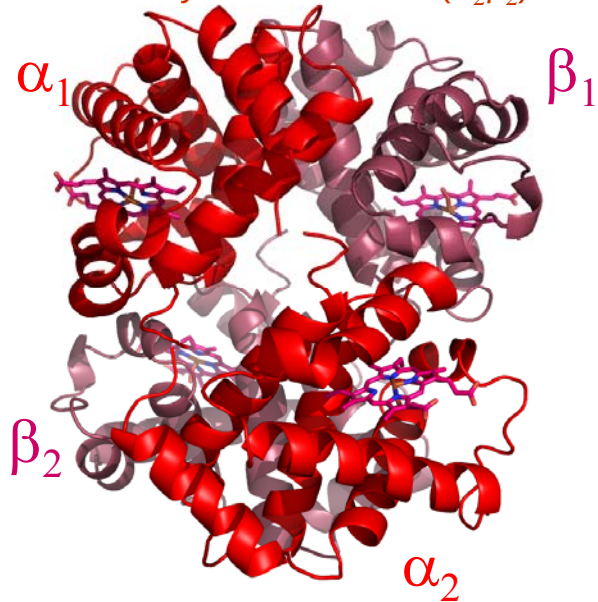
Conformation of deoxyHb 4° structure($\alpha_2\beta_2$)



from PDB2hho

52
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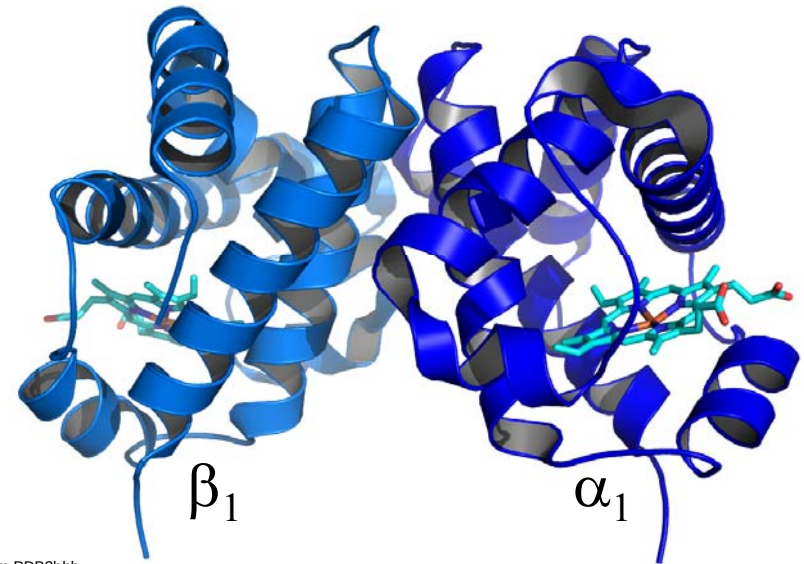
Conformation of oxyHb 4° structure ($\alpha_2\beta_2$)



from PDB1hho

53
© S. Ensign

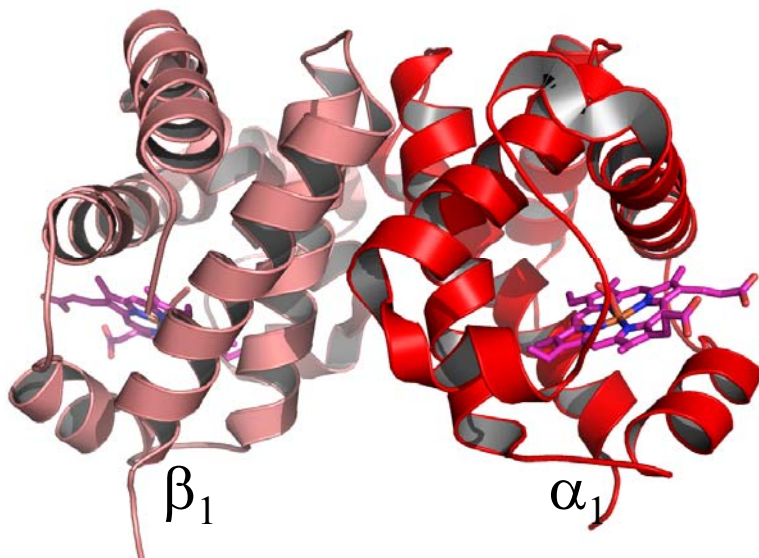
Conformation of deoxyHb ($\alpha_1\beta_1$)



from PDB2hhb

54
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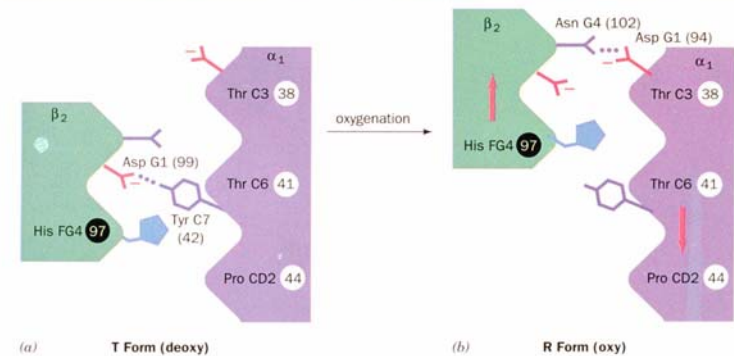
Conformation of oxyHb ($\alpha_1\beta_1$)



55
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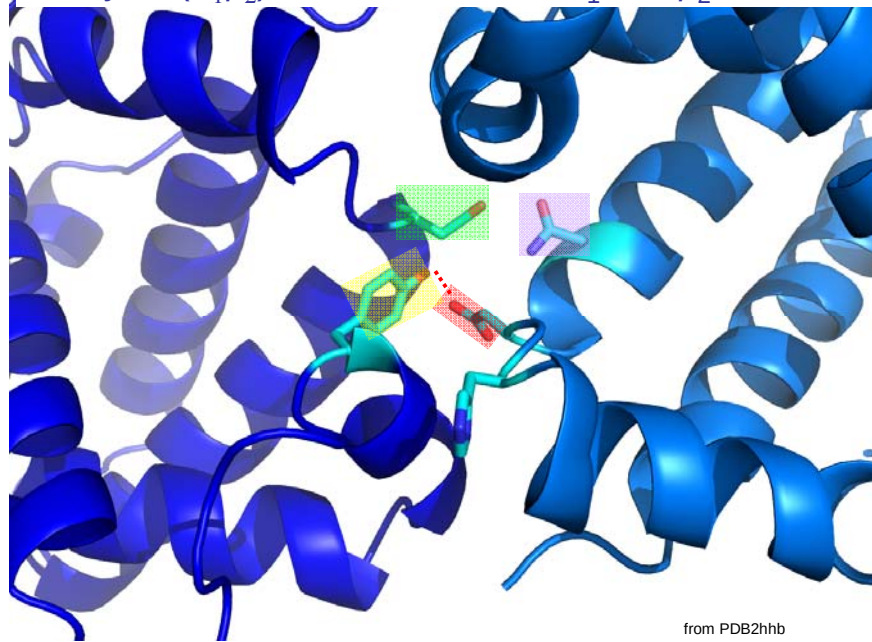
α_1 - β_2 contacts in the T and R states. Identical contacts are present at the α_2 - β_1 interface

Two different sets of H-bonds in 2 states: a switch, allowing only two stable positions of the subunits relative to each other



56
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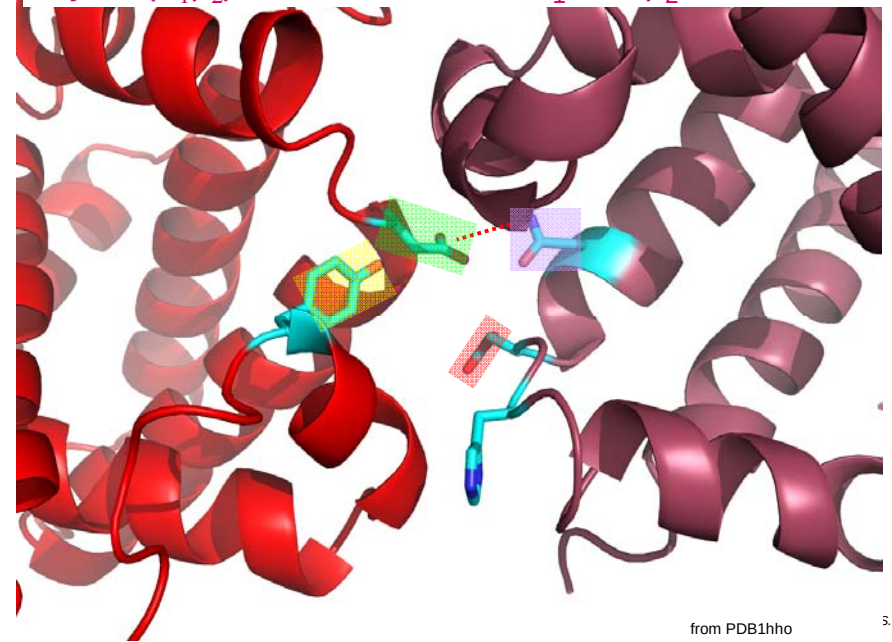
deoxyHb ($\alpha_1\beta_2$) subunit interface: α_1 Y42- β_2 D99 H-bond



from PDB2hbb

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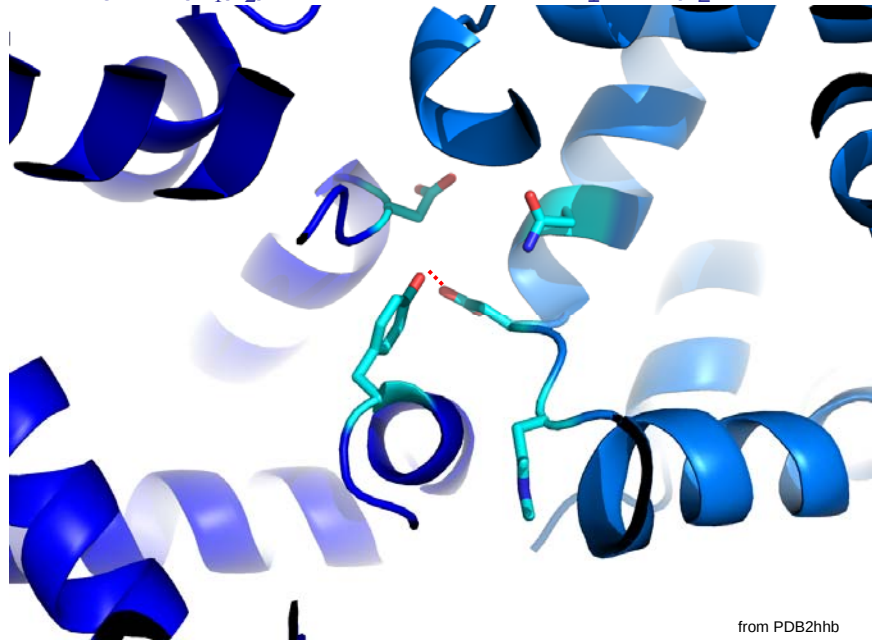
oxyHb ($\alpha_1\beta_2$) subunit interface: α_1 D94- β_2 N102 H-bond



from PDB1hho

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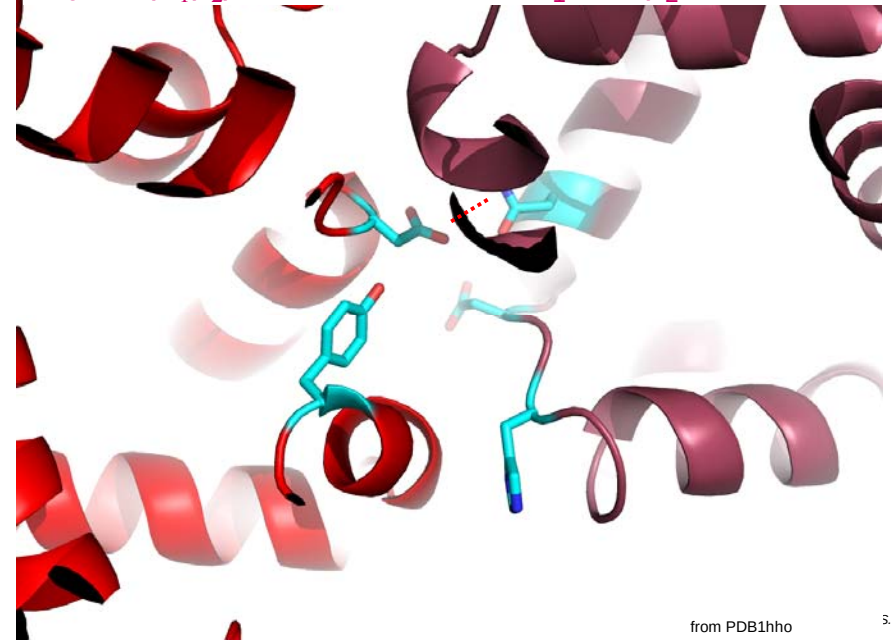
deoxyHb ($\alpha_1\beta_2$) subunit interface: α_1 Y42- β_2 D99 H-bond



from PDB2hbb

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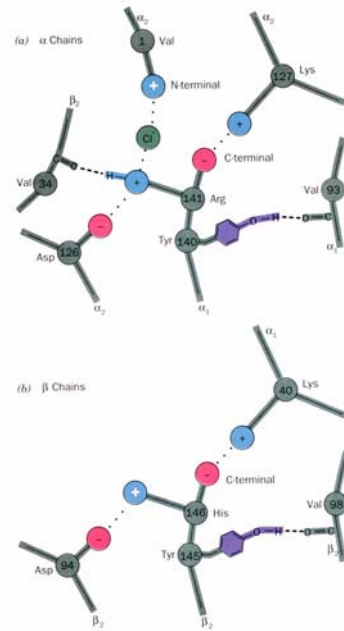
oxyHb ($\alpha_1\beta_2$) subunit interface: α_1 D94- β_2 N102 H-bond



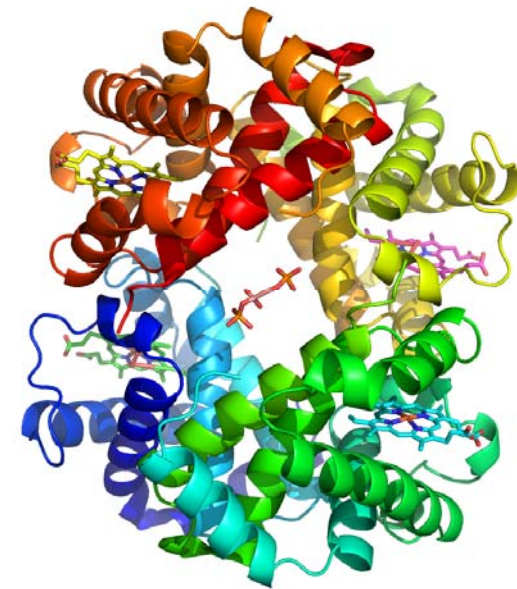
from PDB1hho

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Some representative salt bridges in T Hb

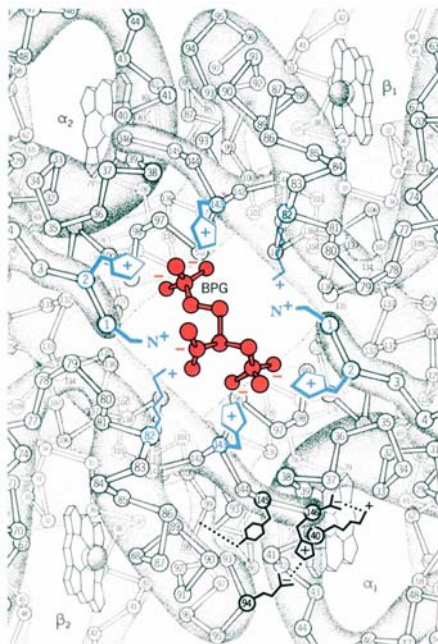


61
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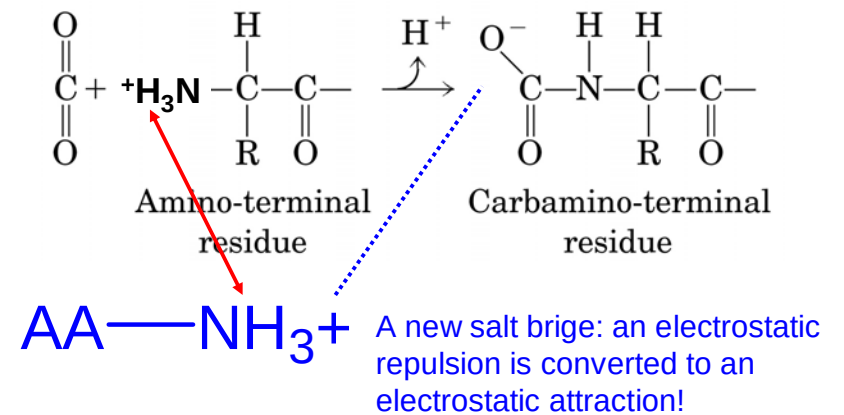
from PDB 1b86

62
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Molecular explanation for the Bohr effect: CO_2



64
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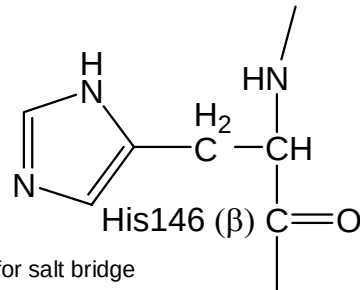
Molecular explanation for the Bohr effect: H^+

R (oxy) hemoglobin in lungs

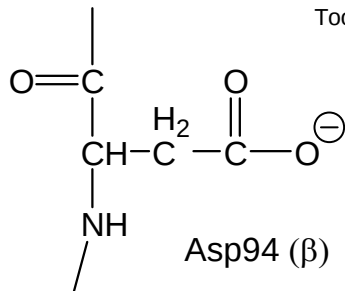
pH in lungs is ~7.4

His146 is deprotonated

(unperturbed His $pK_a = 6$)



Too far apart for salt bridge



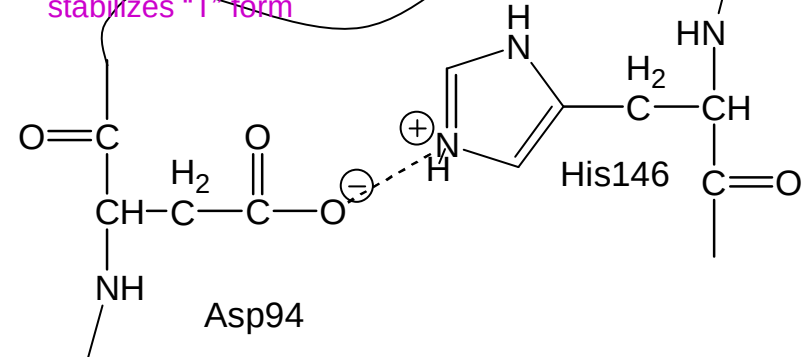
65
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Molecular explanation for the Bohr effect: H^+ T hemoglobin in tissue

pH in tissue is ~7.2

His146 would usually be deprotonated

But closer Asp94 raises its pK_a so it binds a proton and forms a stabilizing intrasubunit salt bridge that further stabilizes "T" form



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Summary of T and R hemoglobin structures

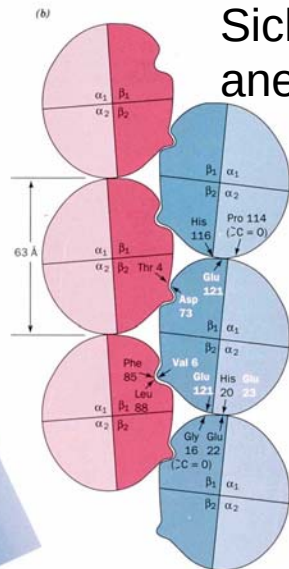
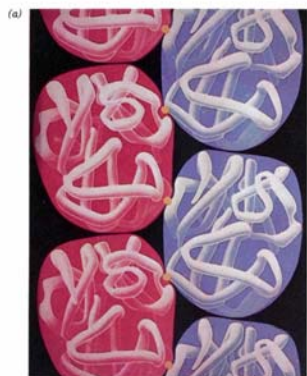
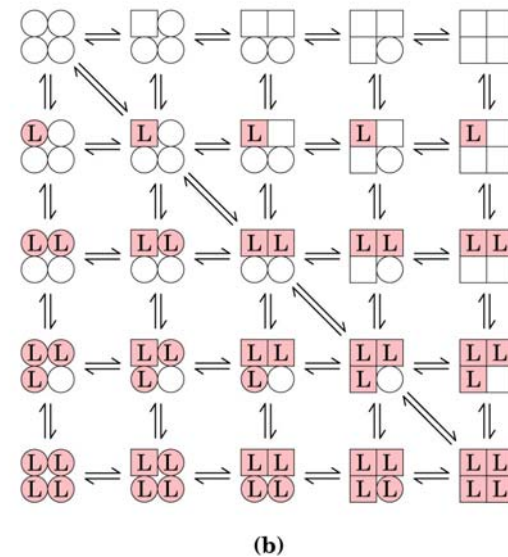
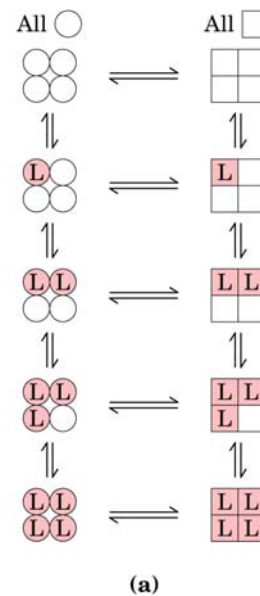
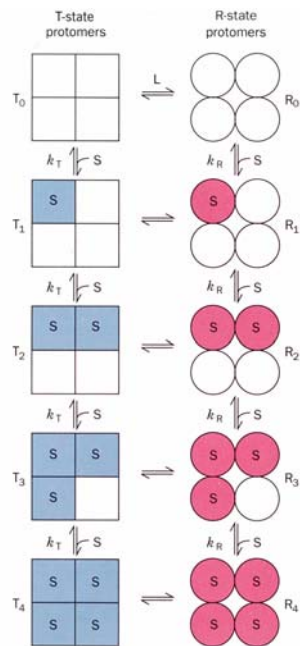
- T Hb (4° structure) is stabilized by multiple salt-bridges, H-bonds, and ion-dipole interactions at subunit-subunit interfaces
- O_2 binding to a "T" Hb subunit induces a conformational change in the heme, reorienting the F helix, which translates to an overall conformational change in the Hb subunit
- In R hemoglobin (all four hemes occupied by O_2) the subunit conformations are all as above, and different subunit-subunit interactions stabilize the 4° structure
- Binding of DPG in the central cavity of T Hb stabilizes the T conformation, decreasing O_2 binding affinity
- Asp94 and His146 on a β subunit are close enough to raise the pK_a of His146 so it can protonate at the lower pH of tissue, forming a stabilizing salt bridge in T Hb
- Carbamaylation of the N-terminal amino group of the α subunits converts a positively charged group to a negatively charged group, allowing formation of a stabilizing salt bridge to a (+) AA in T hemoglobin
- The above information has been deduced based on static structures of Hb without Hb bound (T) or with four O_2 bound per Hb (R)

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Cooperative binding of O_2 to Hb

- O_2 binding to a "T" Hb subunit induces a conformational change in the heme, reorienting the F helix, which translates to an overall conformational change in the Hb subunit
- Presumably, this conformational change in one subunit reorients adjacent subunit(s) making it easier for O_2 to bind successive subunits
- Does the reorientation of the other subunit(s) involve immediate conformational changes (before O_2 binds to them) or does it just make O_2 bind with higher affinity (lower K_d) followed by T to R conformational changes in the subsequent subunits(s)?
- The conformational changes, and overall structures of Hb with 1-3 O_2 bound (induced by sequential O_2 binding) may be different than the structures of "all T" and "all R" Hb
- The mechanism of O_2 -induced cooperativity is not fully understood, but binding of successive O_2 molecules increases affinity for binding of successive O_2
- affinity $O_{2(1)} < O_{2(2)}, O_{2(3)} < O_{2(4)}$

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Sickle cell anemia

